



WWTP modelling



Hydraulic model

- Deduced from mass balances
- Depends on each WWTP
- Kinetic model
 - Rate equations
 - Common for similar processes

The most common kinetic models have been developed by workgroups of the **International Water Association (IWA)** and are known as **Activated Sludge Models (ASM)**. The most used models are:

ASM1	COD + N
ASM2d	COD+ N + P
ASM3	COD + N with COD accumulation by heterotrophic organisms
ADM1	Anaerobic digestion of COD

These models take into account different types of microorganisms and numerous substrates and products. Usually they are described using a matrix notation.



	.				Activated sludge:
				inetic parameters of ASM2d	
		315 of pt 3.00	nticulate substrate: X _S d ⁻¹	** 1 1 *	100 plus 1 years
	K _H	0.60	a	Hydrolysis rate constant	UNIVERSITÀ
	n _{NO3}	0.40		Anoxic hydrolysis reduction factor Anaerobic hydrolysis reduction factor	New trends and
	n _{fe} Koz	0.20	$z O_2/m^3$	Saturation/inhibition coefficient for oxygen	• perspectives
	K _{N02}	0.20	g N/m ³	Saturation/inhibition coefficient for nitrate	
	K _{NO3} K _X	0.10	g X _S (gX _H) ⁻¹	Saturation/inition coefficient for nitrate	
			rganisms: XR	Saturation coefficient for particulate COD	
		6.00	g X _S (gX _R) ⁻¹ d ⁻¹	Maximal growth rate on substrate	
	u _H	3.00	g X _S (gX _H) ⁻¹ d ⁻¹	Maximal growth rate on substrate Maximal fermentation rate	
	Qfe	0.80	B Y2 (BYH) g	Reduction factor for denitrification	
	n _{NO3}	0.40	d-1	Lysis rate constant	
	b _H K ₀₂	0.20	g O ₂ /m ³	Saturation/inhibition coefficient for oxygen	
	K ₀₂ K _F	4.00	g COD m ⁻³	Saturation coefficient for growth on SF	
	K _F	4.00	g COD m ⁻³	Saturation coefficient for growin on SF	
	K _{fe} K _A	4.00	g COD m ³	Saturation coefficient for acetate	
	K _{NO3}		gCODm gNm ⁻³	Saturation coefficient for acetate Saturation/inhibition coefficient for nitrate	
	K _{N03} K _{NH4}	0.50	gNm ⁻ zNm ⁻³	Saturation/inhibition coefficient for nitrate Saturation coefficient for SNH4 as nutrient	
				Saturation coefficient for SPO4 as nutrient	
	Kp	0.01	g P m ⁻³		
	KALK		mole HCO ³⁻ m ⁻³	Saturation coefficient for alkalinity	
			umulating organisms: X _{PAO}	D	
	Q PHA	3.00	(g X _{PHA} (g X _{PAO}) ⁻¹ d ⁻¹	Rate constant for storage of XPHA	
	Q PP	1.50	(g X _{PHA} (g X _{PAO}) ⁻¹ d ⁻¹ d ⁻¹	Rate constant for storage of XPP	
	uPAO	1.00	a '	Maximum growth rate of XPAO Reduction factor under anoxic conditions	
	n _{NO3}	0.60	d ⁻¹		
	b PAO	0.20	a d ⁻¹	Rate for lysis of XPAO	
	bpp	0.20	d-1	Rate for lysis of XPP	
	b _{PHA}	0.20		Rate for lysis of XPHA	
	K ₀₂ K _{N03}	0.50	g O ₂ m ⁻³ g N m ⁻³	Saturation coefficient for oxygen Saturation coefficient for nitrate	
	K _{NO3}	4.00	g COD m ⁻³	Saturation coefficient for intrate	
	K _{NH4}	0.05	gNm ³	Saturation coefficient for acetate	
	K _{NH4} K _P	0.05	g P m ⁻³	Saturation coefficient for ammonium Saturation coefficient for phosphate for XPP formation	
	K _{PO4}	0.20	g P m ⁻³	Saturation coefficient for phosphate for APP formation Saturation coefficient for phosphate for growth	
	K _{ALK}	0.10	mole HCO ³⁻ m ⁻³	Saturation coefficient for phosphate for growin Saturation coefficient for alkalinity	
	K _{ALK} K _{pp}	0.10	g Xpp(g XpAQ) ⁻¹	Saturation coefficient for alkalinity Saturation coefficient for polyphosphate	
	KMAX	0.34	g X _{PP} (g X _{PAO})-1	Maximum ratio of XPP/XPAO	
	KIPP	0.02	g Xpp(g XpAO)-1 g Xpp(g XpAO)-1	Inhibition coefficient for polyphosphate storage	
	K _{PHA}	0.02	g Xpp(g XpAO)-1 g Xpha(g XpAO)-1	Saturation coefficient for PHA	
			isms (autotrophic organisms)		
		1 <u>g organ</u> 1.00	d-1 d-1	Maximal growth rate of autotrophic biomass	
	u _{AUT}	0.15	d-1	Decay rate if autotrophic biomass	
	K ₀₂		$g O_2 m^{-3}$	Saturation/inhibition coefficient for oxygen	
	K ₀₂ K _{NH4}	1.00	g N m ⁻³	Saturation/infibition coefficient for oxygen Saturation coefficient for SNH4	
	K _{ALK}	0.50	mole HCO ³⁻ m ⁻³	Saturation coefficient for SIG14	
	KALK		g P m ⁻³	Saturation coefficient for SPO4	
			horus removal	Saturation coefficient for 5PO4	
		<u>ai phosp</u> 1.00	m ³ (gFe(OH) ₃) ⁻¹ d ⁻¹	Pote constant for Doministration	
	k _{PRE}	0.60	m ⁻ (gFe(OH) ₃) ⁻ d ⁻¹	Rate constant for P precipitation Rate constant for redissolution	
	k _{RED}		nole HCO ³⁻ m ⁻³	Saturation coefficient for alkalinity	
	K _{ALK}	0.50	more rico m	Saturation coefficient for analimity	
Biological nutrient remo	val: ma	them	atical modelling as a	a good strategy for control system design	(J.A. Baeza)



																UNIVER DECLI ST DI PALER	-) plus	New tre	-
	Table A3. Stoichiom	etry matrix f	ior ASM	E2d (v _{ij})															
j	i: Process	S ₀₂	S ₇	S _A	S _{NH4}	S _{NO8}	S204	\$ ₁	SALK	Sm	X	Xs	Xg	XPAO	Xn	XPRA	XAUT	X _{TSS}	X _{M+OH}	Хма
	Aerobic hydrolysis	1	$1-f_{\rm SI}$		V _{1,NH4}		v _{1,P04}	\mathbf{f}_{SI}	VLALK			-1						V _{1,TSS}		
	Anoxic hydrolysis	1	1-f _{st}		V2,NH4		V2,P04	fst	V2,ALK			-1						V2,TSS		
	Anaerobic hydrolysis	1	1-f _{st}		V _{3,NH4}		v _{3,PO4}	\mathbf{f}_{SI}	V _{3,ALK}			-1						v _{3,TSS}		
	Aerobic growth on S_F	$1 - \frac{1}{Y_H}$	$-\frac{1}{Y_H}$,			-IP,BM						1							
	Aerobic growth on ${\rm S}_{\rm A}$	$1 - \frac{1}{Y_H}$		$-\frac{1}{Y_H}$			-ір,вм						1							
	Anoxic growth on $S_{\rm F}$, denitrification		$-\frac{1}{Y_{H}}$			$-\frac{1-Y_H}{2.86 \cdot Y_H}$	-i _{P,BM}			$\frac{1-Y_H}{2.86 \cdot Y_H}$			1							
	Anoxic growth on S_A , denitrification			$-\frac{1}{Y_H}$		$-\frac{1-Y_H}{2.86 \cdot Y_H}$	-i _{P,BM}			$\frac{1-Y_H}{2.86 \cdot Y_H}$			1							
	Fermentation		-1	1																
	Lysis										\mathbf{f}_{NI}	$1-f_{33}$	-1							
)	Storage of X _{PHA}			-1			Y _{PO4}								-Y _{P04}	1				
L	Aerobic storage of X _{PP}	-Y _{PHA}					-1								1	-Y _{PHA}				
2	Anoxic storage of X _{PP}					V12,N03	-1			-V12,N03					1	-YPHA				
3	Aerobic growth X _{PAO}	$v_{13,02}$					-i _{P,BM}							1		$-\frac{1}{Y_{PAO}}$				
	Anoxic growth X_{PAO}					V14,N03	-i _{P,BM}			-V14,N03				1		$-\frac{1}{Y_{PAO}}$				
	Lysis of XPAO						V15,P04				fx	1-f _M		-1		* PAO				
5	Lysis of Xpp						1								-1					
,	Lysis of X _{PHA}			1												-1				
8	Aerobic growth of X_A	$-\frac{4.57-Y_A}{Y_A}$			$-i_{N,BM} - \frac{1}{T_A}$	$\frac{1}{Y_A}$	-i _{P,BM}		VIRALK								1			
,	Lysis of X _A				V19,NH4		V19,P04				\mathbf{f}_{NI}	$1-f_{\rm NI}$					-1			
)	Precipitation						-1		V _{20,ALK}									1.42	-3.45	4.87
l	Redissolution						1		V21,ALK									-1.42	3.45	-4.87



	wv	VTP n		SM2d calibrat		Activated sludge 100 plus 1 years New trends and perspectives
Plant o	lata.			ASM2d state v	ariable	es:
Plant	iata:		Symbol	Description	Symbol	Description
0.01			\mathbf{S}_{O2}	Dissolved oxygen concentration, [g O ₂ m ⁻³]	$\mathbf{X}_{\mathbf{S}}$	Slowly biodegradable substrates, [g COD m ⁻³]
COL)		\mathbf{S}_{F}	Readily biodegradable soluble organic substrate, [g COD m ⁻³]	$X_{\rm H}$	Heterotrophic organisms, [g COD m-3]
BOD	5		\mathbf{S}_{A}	Fermentation products VFA, [g COD m ⁻³]	X_{PAO}	Phosphorus accumulating organisms, [g COD m ⁻³]
TKN	1		S_{I}	Inert soluble organic material,[g COD m ⁻³]	\mathbf{X}_{PP}	Polyphosphate, [g P m ⁻³]
NH			$\mathbf{S}_{\mathrm{NH4}}$	Ammonium plus ammonia nitrogen, [g N m ⁻³]	\mathbf{X}_{PHA}	Cell internal storage product of PAO, [g COD m ⁻³]
NO			S_{N2}	Gaseous nitrogen, [g N m ⁻³]	$\mathbf{X}_{\mathrm{AUT}}$	Nitrifying organisms, [g COD m ⁻³]
	•		$\mathbf{S}_{\mathrm{NO3}}$	Nitrate plus nitrite nitrogen, [g N m ⁻³]	\mathbf{X}_{TSS}	Total suspended solids, TSS, [g TSS m ⁻³]
PO, VSS			$\mathbf{S}_{\mathrm{PO4}}$	Inorganic soluble phosphorus, [g P m ⁻³]	$\rm X_{MeOH}$	Metal-hydroxides, involved with chemical removal of phosphorus, [g TSS m ⁻³]
тรร	5		$S_{ALK} \\$	Alkalinity of the wastewater, [mol HCO ₃ m ⁻³]	$\mathbf{X}_{\mathrm{MeP}}$	Metal phosphate, [g TSS m ⁻³]
		_	X _I	Inert particulate organic material, [g COD m ⁻³]		
Biological nut	rient rem	noval: mathema	tical mod	lelling as a good strategy for co	ontrol syst	em design (J.A. Baeza)



	W	/WTP n	nodelling		otivated sludg
			ASM2d calibration	on	
	<i>S_{i j}</i> =	$=\frac{\theta_j}{y_i}\frac{dy_i}{d\theta_j}$	Selection of pa Sensitivity analysis/		fit
0		5	$+ \left S_{j,NO_3} \right + \left S_{j,XTSS} \right + \left S_{j,TKN} \right $	$FIM = \sum_{k=1}^{N} Y_{\theta}(k)$	$\cdot Q_k^{-1} \cdot Y_{\theta}^T ($
			Kinetic / Stoichiometric Group (K	group)	
	Order	Parameter	Short Description	Related biomass or process	Sensitivit
	1	Y _H	Yield coefficient for X _H .	Heterotrophic	756
	2	μ_{A}	Maximum growth rate of X _A	Autotrophic	678
	3	b _A	Rate for lysis of X _A	Autotrophic	634
	3 4	b _A K _{NH4,A}	Rate for lysis of X_A Saturation coefficient of substrate NH_4^+ for nitrification on S_{NH4}	Autotrophic	634 412
			Saturation coefficient of substrate	•	
	4	K _{NH4,A}	Saturation coefficient of substrate NH_4^+ for nitrification on S_{NH4}	Autotrophic Chemical phosphate	412
	4	K _{NH4,A} K _{PRE}	Saturation coefficient of substrate NH_4^* for nitrification on S_{NH4} Precipitation constant Saturation coefficient of O_2	Autotrophic Chemical phosphate precipitation Autotrophic Chemical phosphate	412 150
	4 5 6	K _{NH4,A} K _{PRE} K _{02,A}	$\begin{array}{l} \text{Saturation coefficient of substrate} \\ \text{NH}_4^* \text{ for nitrification on S_{NH4}} \\ \text{Precipitation constant} \\ \text{Saturation coefficient of O_2} \\ \text{for nitrification on S_{NH4}} \end{array}$	Autotrophic Chemical phosphate precipitation Autotrophic	412 150 149
	4 5 6 7	K _{NH4,A} K _{PRE} K _{O2,A} K _{RED}	Saturation coefficient of substrate NH_4^* for nitrification on S_{NH4} Precipitation constant Saturation coefficient of O_2 for nitrification on S_{NH4} Solubilisation constant	Autotrophic Chemical phosphate precipitation Autotrophic Chemical phosphate precipitation	412 150 149 148



WWTP modelling



After a proper process of **model calibration** and **validation**, we can be confident that we have a model able to provide a **good description of** the simulated variables in **that particular WWTP**

However, we need to calculate different **performance indicators** as a tool for comparison of the behaviour of the WWTP under several **operating conditions** or **control strategies**























Optimization of setpoints allows to obtain the better performance of a CS

Minimization of a Cost Function where all the criteria considered are converted to monetary units.

ANALYSED SCENARIOS

- Open Loop (OL): TSS control loop in R4. Aeration constant in R3 and R4. $Q_{RINT}/Q_I = 3$, $Q_{REXT}/Q_I = 1$.
- DO control (DOC): DO control was activated with a setpoint of 4 mg DO $L^{\text{-1}}$ in R3 and R4.
- Maximum performance for nutrient removal (MPR): Ammonia setpoint was 0 mg L⁻¹ and nitrate setpoint was optimised to minimise nitrate in the effluent.
- Ammonium and nitrate fixed optimum setpoints (A&N-FOS): Fixed optimum ammonium and nitrate setpoints.
- Ammonium and nitrate daily variable optimum setpoints (A&N-DVOS): Setpoints daily optimised according to the influent flow pattern of the plant.
- Ammonium and nitrate weekly variable optimum setpoints (A&N-WVOS): Two different sets of setpoints are optimised, one for weekend and one for the weekdays.
- Ammonium and nitrate hourly variable optimum setpoints (A&N-HVOS): Setpoints are hourly optimised according to the influent flow pattern of the plant.









Conclusions of the multi-criteria study

- Multi-criteria optimisation provides a set of optimal operation setpoints approximated by a Pareto surface. The optimised setpoint within this surface can be selected by the requirements that are established for each WWTP in terms of the three criteria.
- These requirements can be translated into monetary weights as was done with OCF. OCF optimisation results in an optimised scenario located on Pareto surface.
- The approaches of single OCF or multi-criteria are complementary. The multi-criteria function enabled a more extensive evaluation of different alternatives where none of the criterion is conditional to the other. Once the weights are selected according to the WWTP requirements, the OCF optimisation could be used to adapt the plant operation to the influent variations.





Selection of CS based on classical control tools

Relative Gain Array (RGA)

Tool for selecting decentralized control structures



Information 1: Best pairing

Information 2: Best set of variables

60	ntrol		ł	landle Va	riables	
		Ма	nipulated	Variables	(MV)	Disturbance (DV)
M	atrix	MV1	MV2	MV3	MV4	DV1
ŝ	CV1	1.00	0.00	0.00	0.00	
s (CV)	CV2	0.00	1.00	0.00	0.00	
Controlled Variables (C	CV3	0.00	0.00	1.00	0.00	
Cc	CV4	0.00	0.00	0.00	1.00	











Cascade + Override Phosphorus Control Strategy (COPCS)









Activated sludge: 100 *plus* 1 years (1-1)













Modelling and equipment limitations ... and consequences in control design

- Analysers precision is limited.
 - Several setpoints reported in the literature cannot be achieved in practice. A sensitivity analysis of optimised controllers setpoint should be performed.
- Manipulated variables have a limited range of operation.
 - · Limits must be considered for all the equipment.
 - Anti wind-up controllers should be used.
- Optimization of controller setpoint provides better improvements than the perfect tuning
 of the controller.
- Sensors dynamics should be considered, but the WWTP dynamics is usually much slower.
- Only water line is usually modelled for control. Internal P inputs from reject water and other recycle streams should be also considered.







Application of Non-Square Relative Gain Array (NSRGA) to WWTP with EBPR

Outputs	Inputs		
	QCOD	Q _{RINT}	Q _{REXT}
Spo4 R2	$\frac{4.935}{0.286s+1}e^{-0.312s}$	$\frac{-2.187 \times 10^{-6}}{0.171s+1} e^{-0.312s}$	$\frac{-1.289 \times 10^{-3}}{0.417s+1} e^{-0.286s}$
S _{PO4} R4	$\frac{-0.266}{1.028s}e^{-0.312s}$	$\frac{-2.231 \times 10^{-5}}{1.091s+1} e^{-0.202s}$	$\frac{1.022 \times 10^{-4}}{0.668s+1} e^{-0.312s}$
S _{PO4} R7	$\frac{-0.992}{0.748s+1}e^{-0.036s}$	$\frac{-2.404 \times 10^{-6}}{0.864s+1} e^{-0.312s}$	$\frac{1.577 \times 10^{-4}}{0.958s+1} e^{-0.307s}$
S _{NO3} R4	$\frac{-0.783}{0.486s+1}e^{-0.069s}$	$\frac{-8.203 \times 10^{-5}}{0.588s+1} e^{-0.307s}$	$\frac{-5.693 \times 10^{-5}}{0.161s+1} e^{-0.312s}$
S _{NO3} R7	$\frac{-1.010}{0.447s+1}e^{-0.295s}$	$\frac{3.24\times10^{-6}}{0.539s+1}e^{-0.295s}$	$\frac{-1.357 \times 10^{-4}}{0.206s+1}e^{-0.307s}$
S _{NH4} R7	$\frac{-1.010}{0.447s+1}e^{-0.295s}$	$\frac{1.235 \times 10^{-6}}{8.210s+1} e^{-0.239s}$	$\frac{-1.072 \times 10^{-6}}{0.840s+1} e^{-0.291s}$
M		de REMARIAN	Computers and Chemical Engineering 53 (2013) 164-177.
INTERNAL	®		Computers and Chemical Engineering
N RECYCLE		- Fride Bar	journal homepage; www.elsevier.com/locate/compchemeng
	0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	LEARNING	
		DRUENT	
		Development and e strategies for optim	conomic assessment of different WWTP contro al simultaneous removal of carbon, nitrogen
	actor 6 Reactor 7 6000 m3	Development and e strategies for optim and phosphorus	







Application of Non-Square Relative Gain Array (NSRGA) to WWTP with EBPR

Control Structures

	controller	Kc	Ti [days]	Td [day
d Controlled variable				
S _{PO4} R2	PID	0.4148	0.4432	0.1012
S _{PO4} R2				-
				-
				-
So. R6	PI	100	0.01	-
So., R7	PI	100	0.01	-
	e finite time norizon, a	and for the comput	tation of the se	quence of
ure control moves.				
	vas used for defining t	these controllers		
	vas used for defining t	these controllers.		
	vas used for defining t	these controllers.		
	vas used for defining t	these controllers.		
	vas used for defining t	these controllers.		
	vas used for defining t	these controllers.		
	vas used for defining t	these controllers.		
	Variable Srot, #2 Srot, #2 Srot, #7 So, #6 So, #6 So, #7 Model F C architecture use	Variable Srev, R2 PID Srev, R2 PI Srev, R3 PI Srev, R4 PI Srev, R5 PI Srev, R5 PI Srev, R7 PI Srev, R6 PI Srev, R7 PI Srev, R6 PI Srev, R7 PI Sreve, R7 PI Sreve, R7 PI Sreve, R7 PI Station PI Station PI Station PI Station	variable Sros, R2 PID 0.4148 Sros, R2 PI -1900.3 Sros, R4 PI 34.635 Sros, R5 PI -4.9246 Sros, R6 PI 100 Sros, R7 PI -1.92246 Sros, R6 PI 100 Sros, R7 PI 100 Sros, R8 PI 100 Sros, R7 PI 100	variable \$vo_R R2 PID 0.4148 0.4432 \$vo_R R2 PI -1900.3 0.56 \$wo_R R4 PI 34,635 0.74 Xrs. R7 PI -4.5246 17 \$vo_R S5 PI 100 0.01 \$vo_R S6 PI 100 0.01



Application of Non-Square Relative Gain Array (NSRGA) to WWTP with EBPR

 Table 7

 Control loops and optimal setpoints of the implemented control strategies.

	Controlled parameter	Controller algorithm	Manipulated variable	Manipulated variable constrains	Optimal setpoint [mg/L]
Control loops for CS1	S_{0_2} in R5, R6, R7 S_{NO_3} in R4 S_{PO_4} in R2	PI PI PID	$k_L a$ in R5, R6, R7 Q_{MNT} COD addition	$0-160 d^{-1}$ $0-92,230 m^3/d$ $0-5 m^3 d^{-1}$	[1.11, 1.45, 0.27] 1.98 27.00
Control loops for CS2	S_{0_2} in R5, R6, R7 S_{NO_3} in R4 S_{PO_4} in R2	PI PI PI	k _L a in R5, R6, R7 Q _{RINT} Q _{REXT}	$\begin{array}{c} 0-160d^{-1} \\ 0-92,\!230m^3/d \\ 9223\!-\!27669m^3/d \end{array}$	[1.00, 1.00, 0.25] 2.00 27.00
Control loops for CS3	$S_{\rm NO_3}$ in R7 $S_{\rm NO_3}$ in R4 $S_{\rm PO_4}$ in R2	Supervisory MPC Slave PI PI PID	S_{O_2} SP in R5, R6, R7 $k_L a$ in R5, R6, R7 Q_{BNT} COD addition	1–2 mg/L R5 and R6 0.25–2 mg/L R7 0–160 d ^{–1} 0–92,230 m ³ /d 0–5 m ³ /d	7.00 Imposed by MPC 2.00 27.00
Control loops for CS4	S_{NH_4} in R7 S_{NO_3} in R4 S_{PO_4} in R2	Supervisory MPC Slave PI PI PID	S_{O_2} SP in R5, R6, R7 $k_L a$ in R5, R6, R7 Q_{RNT} COD addition	1–2 mg/L R5 and R6 0.25–2 mg/L R7 0–160 d ^{–1} 0–92,230 m ³ /d 0–5 m ³ d ^{–1}	1.50 Imposed by MPC 1.92 27.00
Common control loops	TSS in R7	PI	Q_W	300-450 m3/d	3850.00
NTLUEN Racker I Sou m3 Reater Sou m3 Reater Reat	CO CO CO CO CO CO CO CO CO CO	Rector 5 Air Alar Alar Alar Alar Alar Alar Alar Ala	(14) (55) (16) (17) (17) (17) (17) (17) (17) (17) (17	EFFLUENT DISPOSAL (J.A. Bae:	70)



Application of Non-Square Relative Gain Array (NSRGA) to WWTP with EBPR

Control Structures

Inf	Control strategy	AE [€/d]	PE [€/d]	EC [€/d]	SP [€/d]	SNH [€/d]	P _{tot} [€/d]	N _{tot} [€/d]	EF [€/d]	OC [€/d]	OC+dTCC [€/d]
	RO	360	39	0	490	169	683	638	1489	2378	2378
	RO+	382	51	181	502	181	432	483	1096	2212	2212
Drv	CS1	349	49	171	522	143	298	512	953	2044	2093
Diy	CS2	329	43	0	507	165	464	586	1215	2094	2143
	CS3	348	48	169	520	141	310	520	971	2057	2115
	CS4	350	47	181	522	77	303	512	891	1992	2050
	RO	360	39	0	460	385	1019	868	2272	3131	3131
	RO+	382	51	181	484	376	722	666	1764	2861	2861
D	CS1	357	52	400	510	378	506	645	1529	2848	2897
Rain	CS2	318	44	0	458	718	835	935	2488	3309	3358
	CS3	381	51	409	508	199	523	650	1371	2720	2778
	CS4	374	50	419	510	162	518	644	1324	2677	2735
	RO	360	39	0	499	448	846	887	2180	3078	3078
	RO+	382	51	181	509	425	592	676	1693	2816	2816
C1	CS1	361	49	232	527	315	466	705	1486	2655	2704
Storm	CS2	336	42	0	524	358	885	873	2116	3018	3067
	CS3	362	48	232	526	292	474	710	1475	2644	2702
	CS4	373	46	255	528	131	484	704	1318	2520	2578



Conclusions of the NSRGA and MPC study

- Four new control approaches for a WWTP with C/N/P removal, with control loops for improving P-removal in addition to the common C/N loops. All the set-points were optimized to ensure optimal performance → the reported results show the highest feasible performance of these control structures with fixed optimized set-points.
- Comparison with all weather influent files with reference operation (open loop except for TSS control) and with optimized reference operation. These results proved that:

(i) Operational costs and effluent quality of the WWTP can be greatly improved using model based optimization of the reference operation. Optimized reference operation improved effluent quality and operational costs by 7%-9%.

(ii) Automatic control of the WWTP can greatly improve the operational costs of the plant, maintain low pollutant effluent concentrations and achieve a more stable performance.

(iii) The Q_{COD} - PO4 R2 control loop (controlled external carbon addition in the first anaerobic reactor) provides a stable EBPR process and produces a better effluent quality.

(iv) Using the external recycle flow as manipulated variable to control PO4 at the end of the anaerobic zone proved to be a good approach only under dry weather conditions. The Q_{REXT} - PO4 R2 control loop did not assure a stable performance under rain and storm conditions.

(v) CS4 was the most efficient in all working conditions, leading to an operational cost reduction of 120,000 D /year for dry weather conditions. CS3 proved to be the second best due to its good performance during rain and storm events.



Biological nutrient removal: mathematical modelling as a good strate